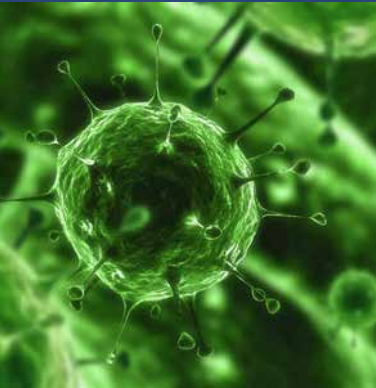


Small-Angle Neutron Scattering: Applications to Multi-Component Systems



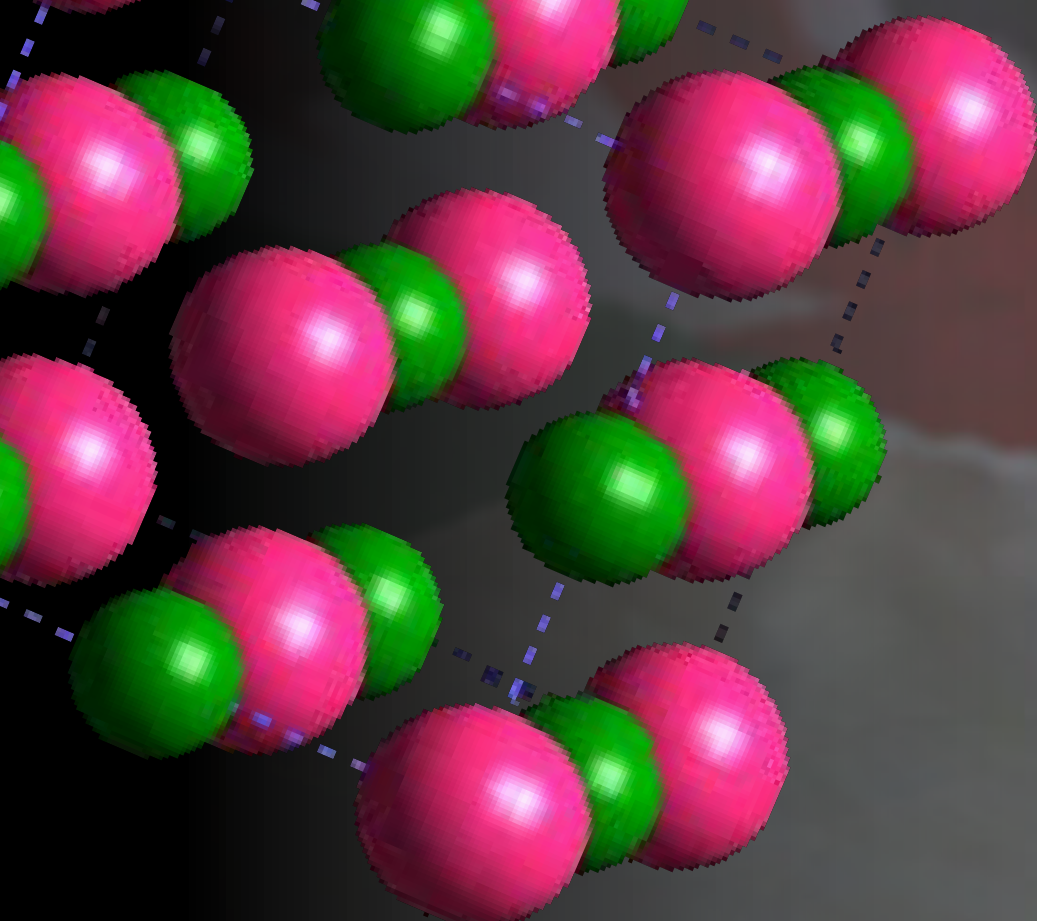
Sarah Rogers

ISIS Facility, Rutherford Appleton Laboratory, UK



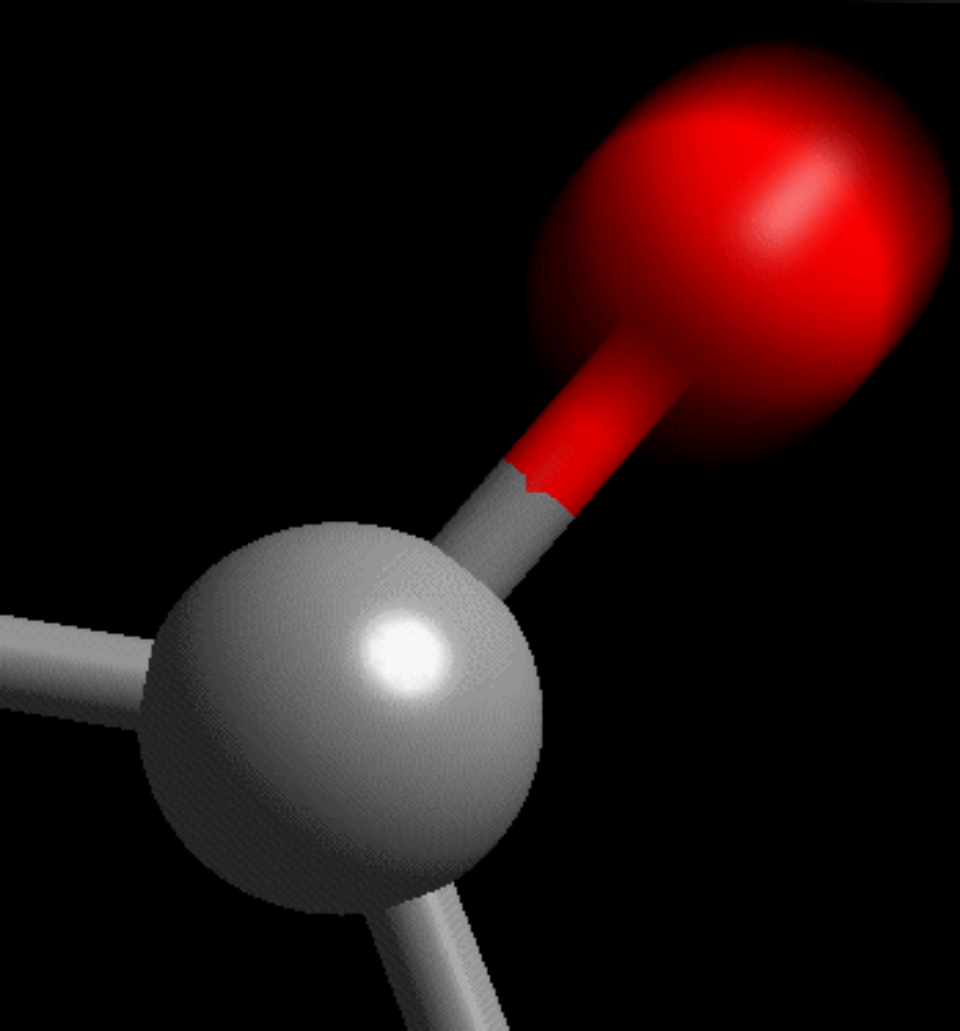
Neutrons for Science





Neutrons

Where atoms are...

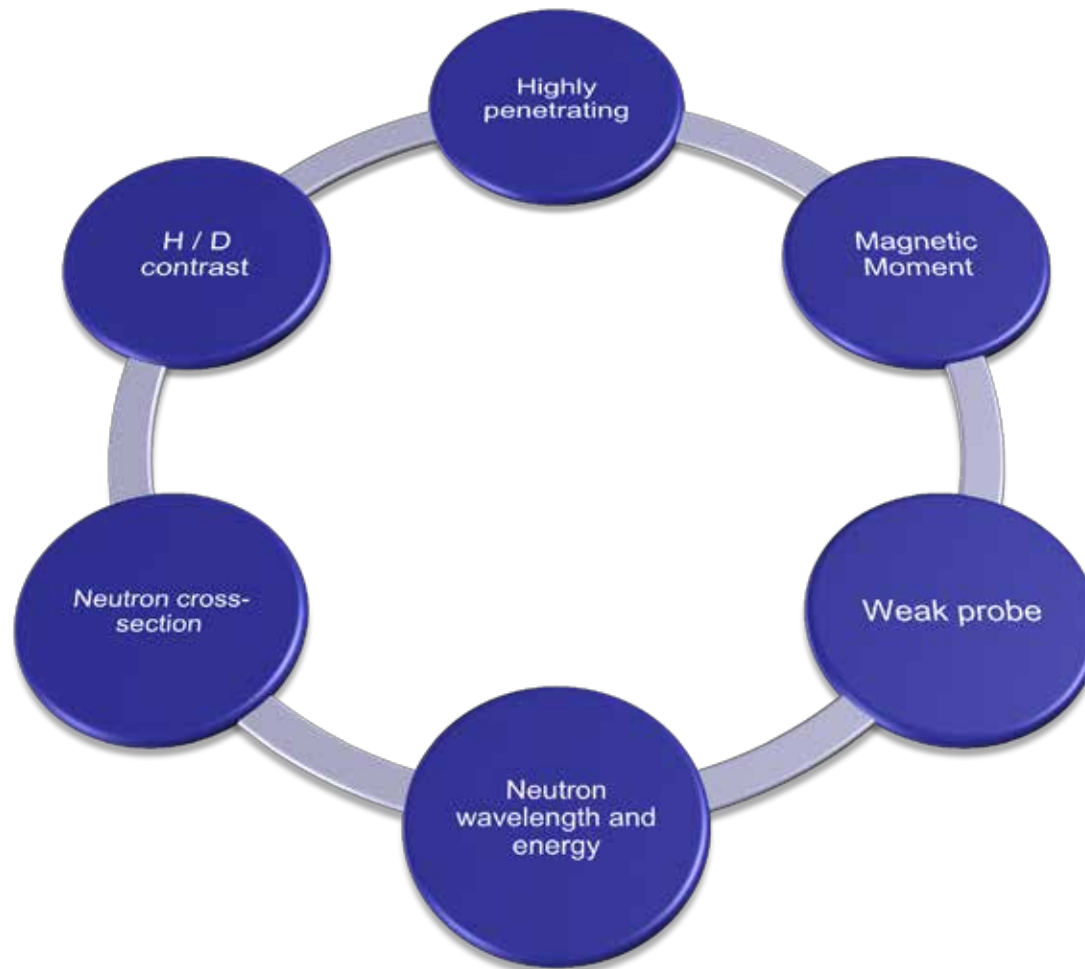


Neutrons

...and what they do

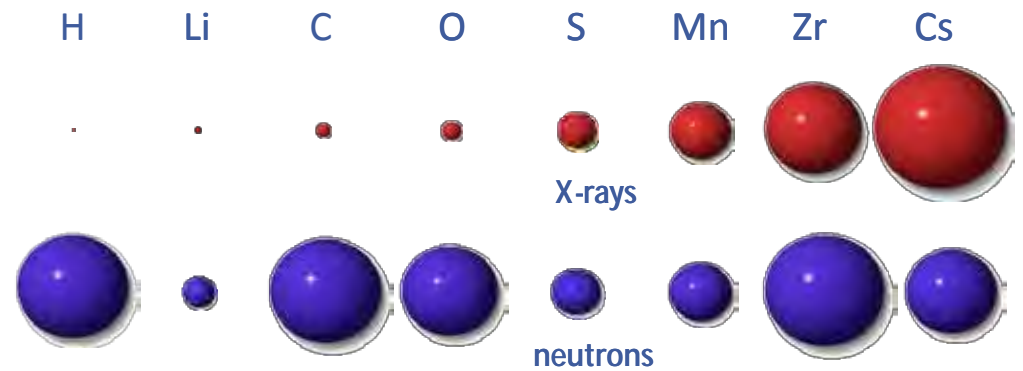
Neutron

Properties



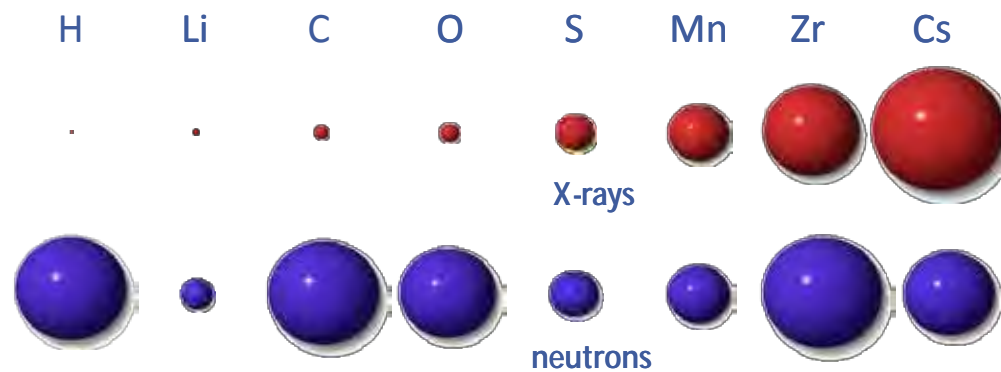
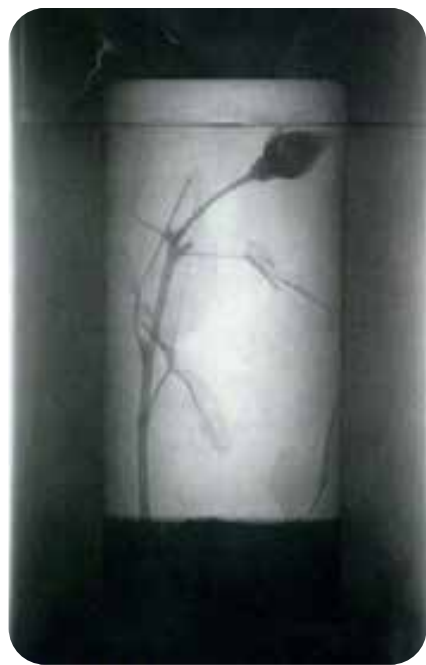
Neutron

Properties

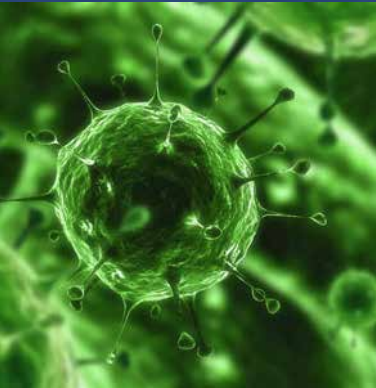


Neutron

Properties



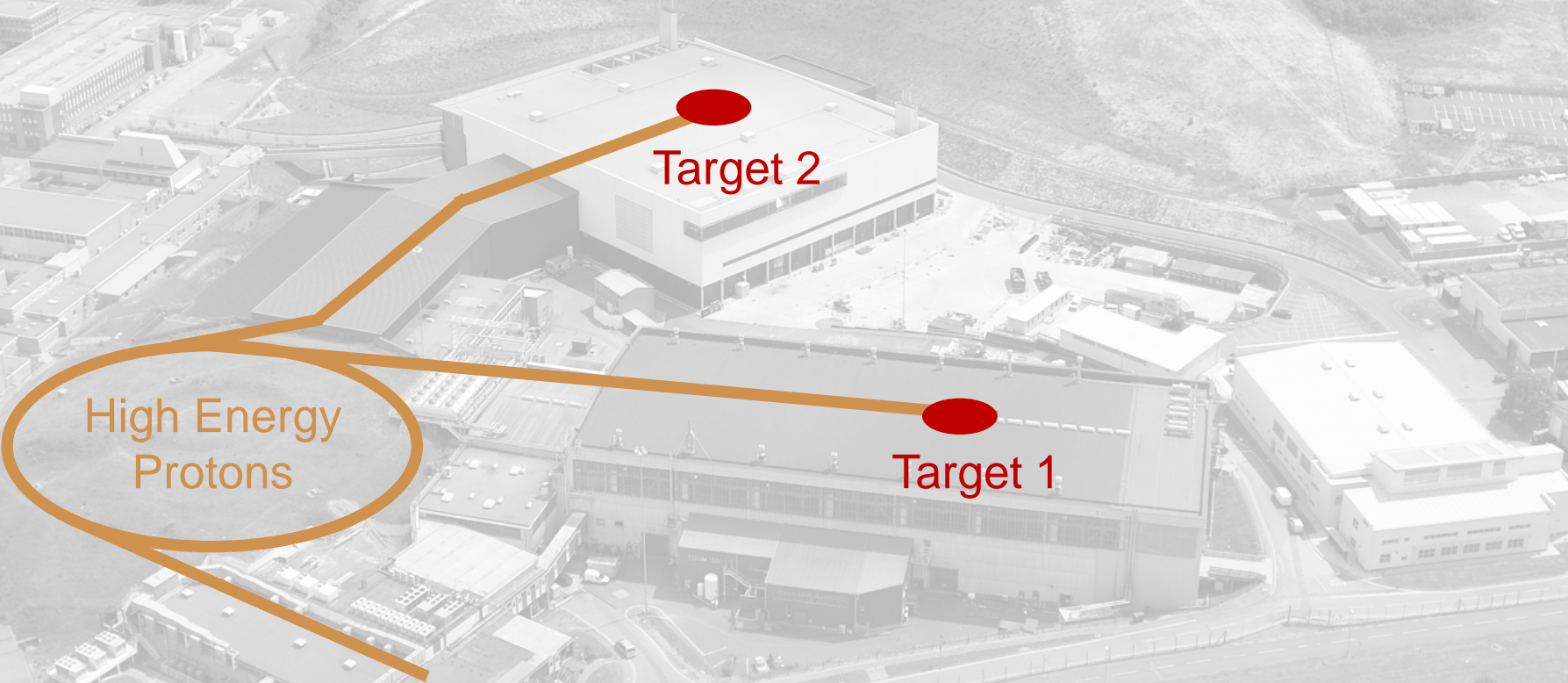
Spallation Neutrons



Accelerator Driven Neutron Source (Spallation)



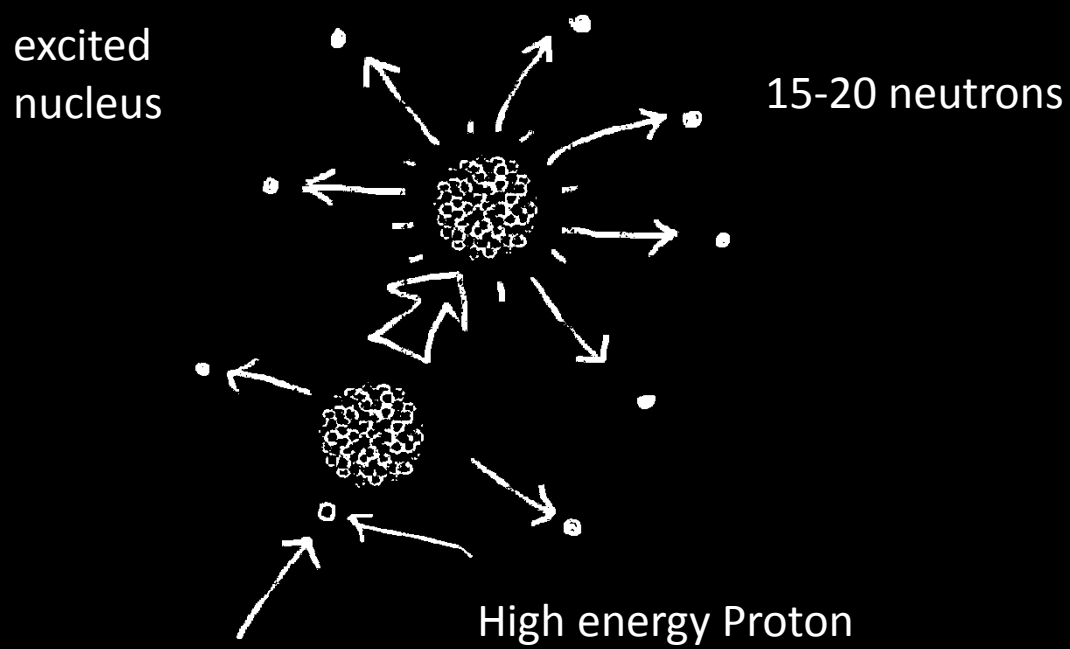
Accelerator Driven Neutron Source (Spallation)



High Energy Protons

Target 2

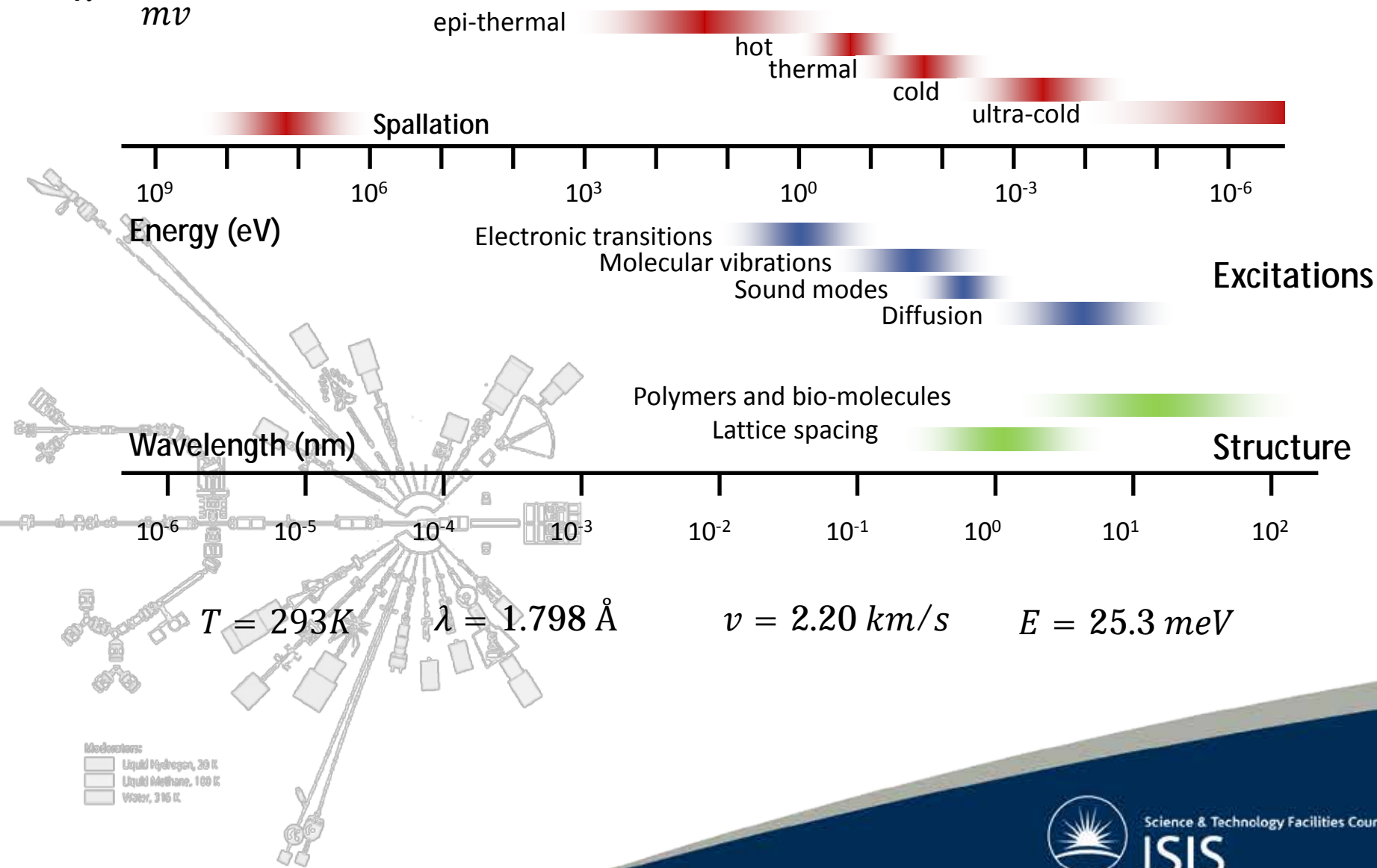
Target 1

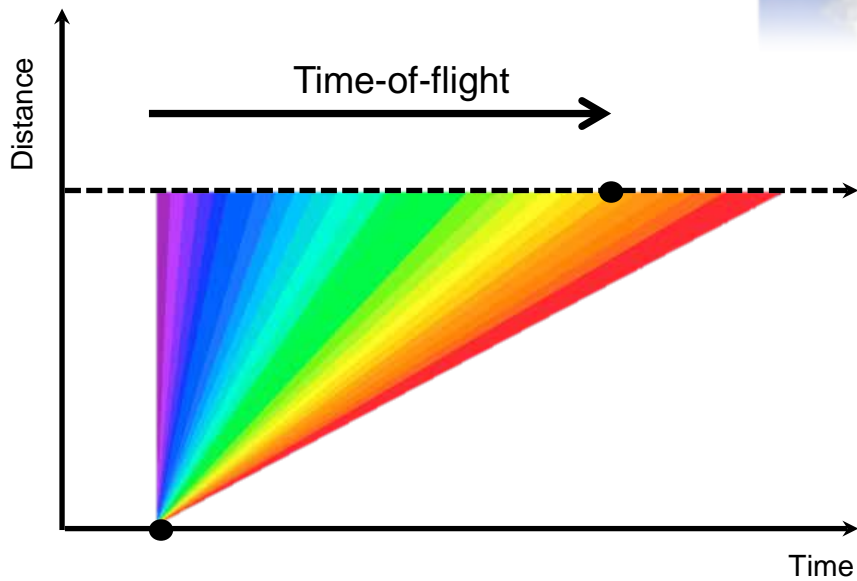
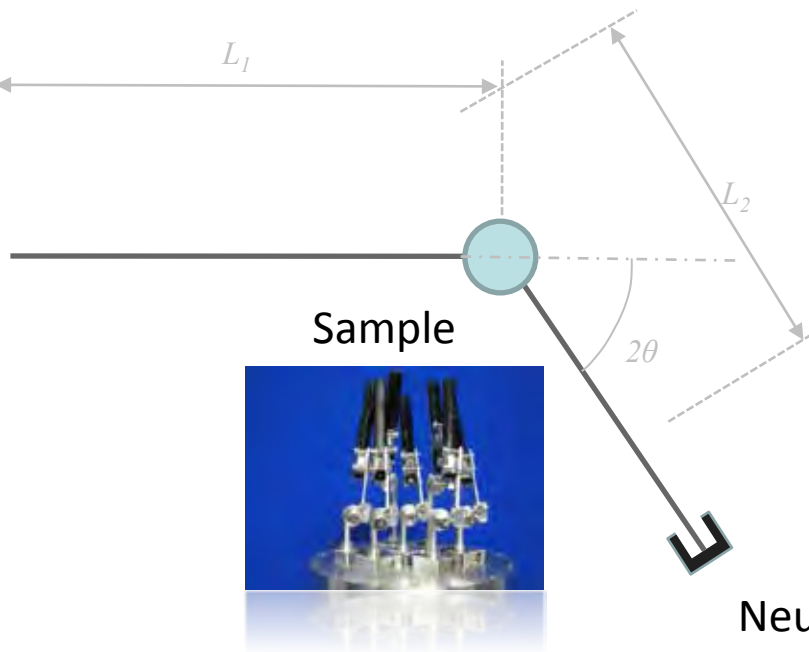


The collision of high energy protons with the tungsten nuclei releases neutrons

$$\lambda = \frac{h}{mv}$$

Moderators





$$v = (L_1 + L_2)/t$$

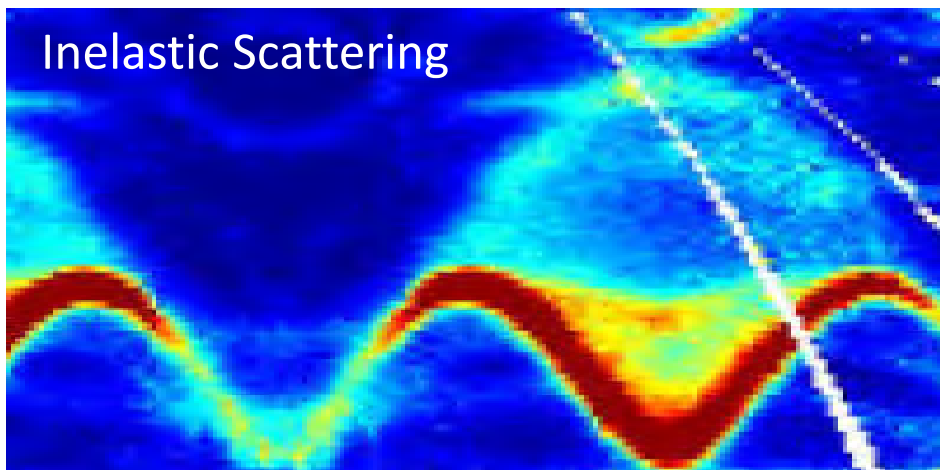
$$E = \frac{1}{2}mv^2$$

$$\lambda = \frac{h}{mv}$$

$$n\lambda = 2d \sin \theta$$



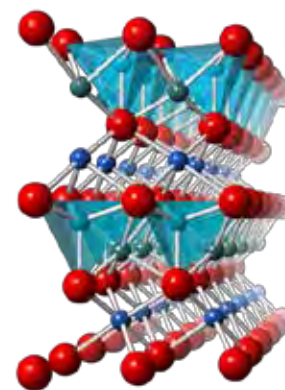
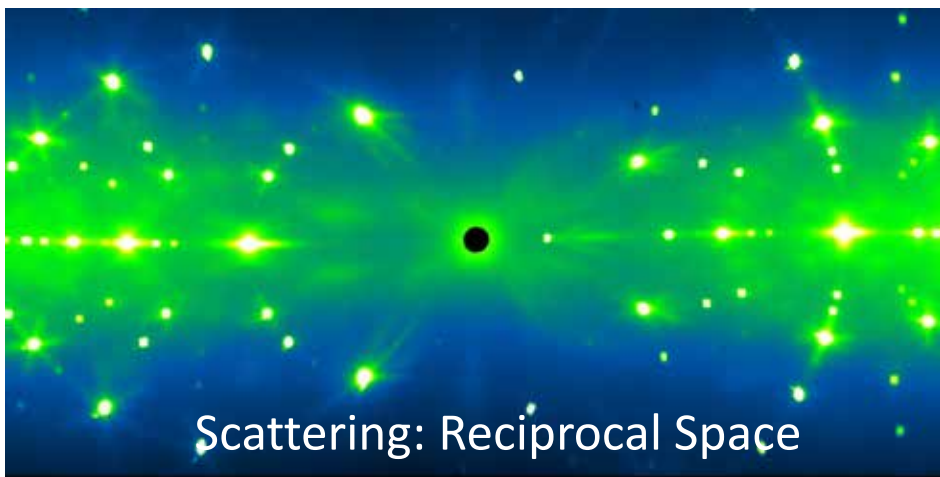
Inelastic Scattering



Dynamical Behaviour



Scattering: Reciprocal Space



Structure: Real Space



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ISIS



Spectroscopy

- MAPS
- LET
- MARI
- MERLIN
- HET

Molecular Spectroscopy

- IRIS
- OSIRIS
- TOSCA
- VESUVIO

Engineering

- ENGIN-X

Large Scale Structures

- SANS2D
- OFFSPEC
- POLREF
- INTER
- CRISP
- SURF
- LOQ

Diffraction

- HRPD
- SXD
- POLARIS
- GEM
- WISH
- PEARL
- INES

30

Neutron and Muons Instruments

Muons

- EMU
- MuSR
- HIFI
- ARGUS

Disordered Materials

- SANDALS
- NIMROD



Target Station 2 – Phase 2 Instruments

- IMAT
- ZOOM
- LARMOR
- CHIPR

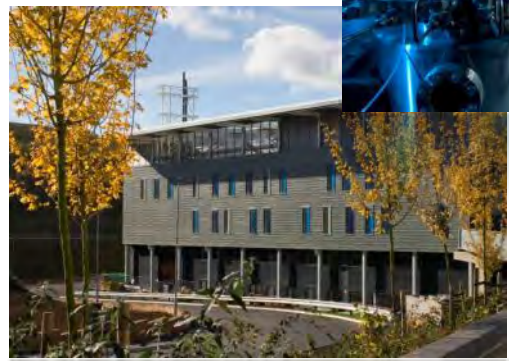
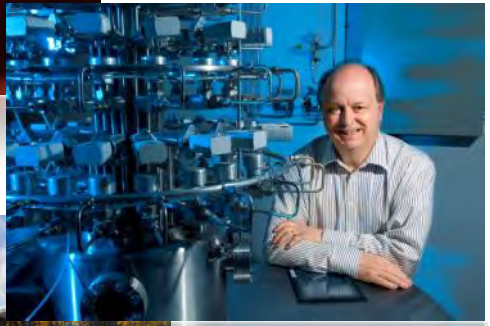
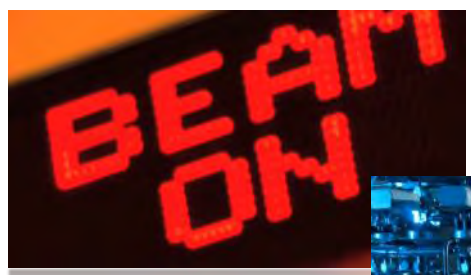




World leading expertise and instrumentation in the application of neutrons to condensed matter science

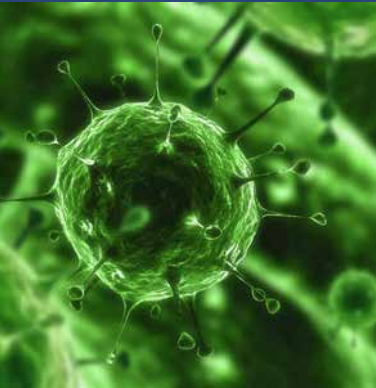


~ 2000 users/yr
~450 publications/yr
~ 800 experiments/yr
90% of UK Users 5/5* Departments



Access Mechanisms
Direct access
Rapid access
Xpress
Programme access
Commercial
ICRD scheme

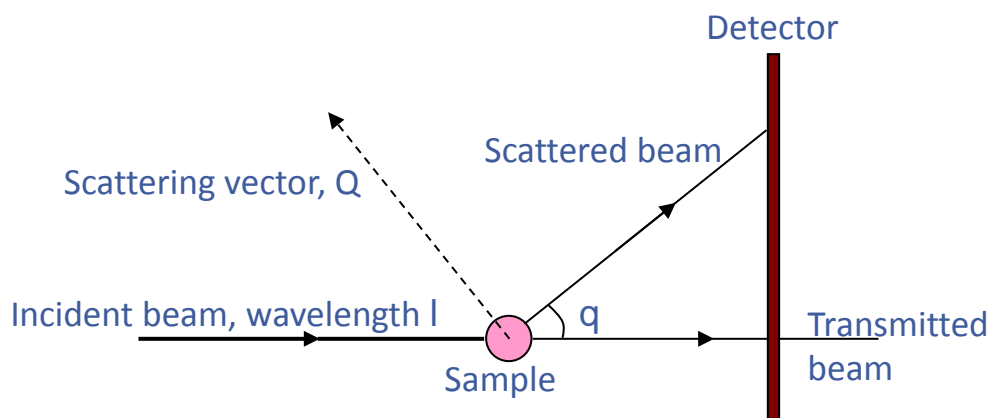
Neutron & Nanometers



Small-Angle Neutron Scattering (SANS)

Lengthscales probed range from 10s to 100s nm. They are explored in reciprocal space by detecting the number of scattered neutrons as a function of the scattering vector, Q . Q is inversely proportional to distance, D , by the approximation:

$$Q = \frac{2\pi}{D}$$



Units are either \AA^{-1} or nm^{-1} i.e. the smaller the value of Q the bigger the object

Q is also related to wavelength and the scattering angle by:

$$Q = \frac{4\pi \sin\left(\frac{\theta}{2}\right)}{\lambda}$$

Q (size) range is varied by altering q or l

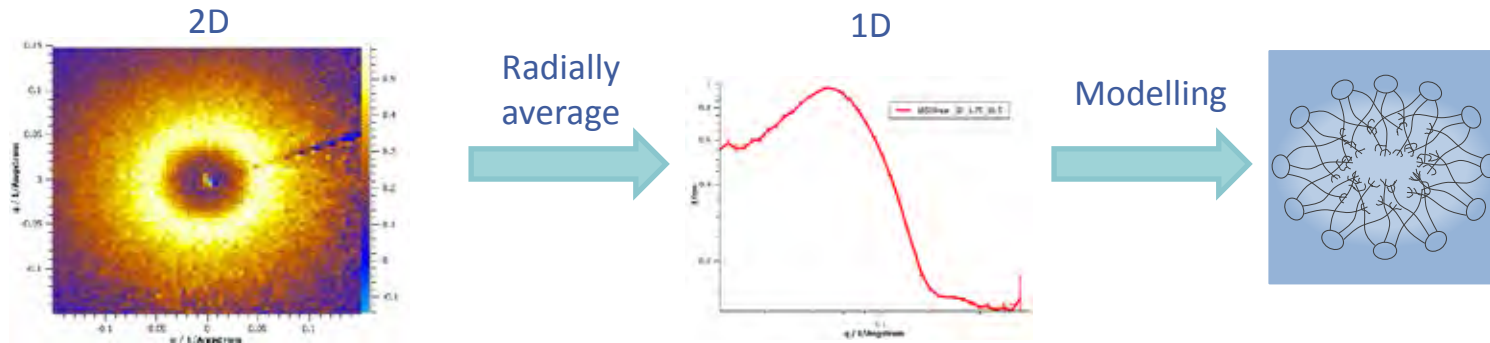
Allows the bulk properties of a material:

- Size
- Polydispersity
- Structure
- Particle Interaction



'Typical' Experiment

The 2D SANS patterns obtained are often radially averaged to give an 'intensity', $I(Q)$, vs. Q plot



$I(Q)$ contains the information on size, shape and interactions between the scattering centres in the sample. For monodisperse spheres $I(Q)$ can be defined as:

$$I(Q) = (\rho_p - \rho_m)^2 N_p V_p^2 P(Q) S(Q) + B$$

Form factor: intra-particle information
- size and shape of particle

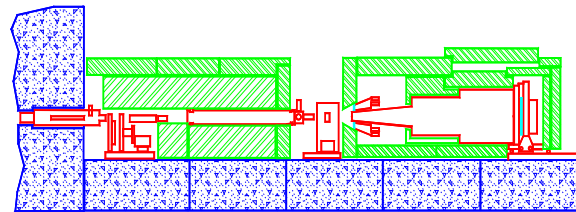
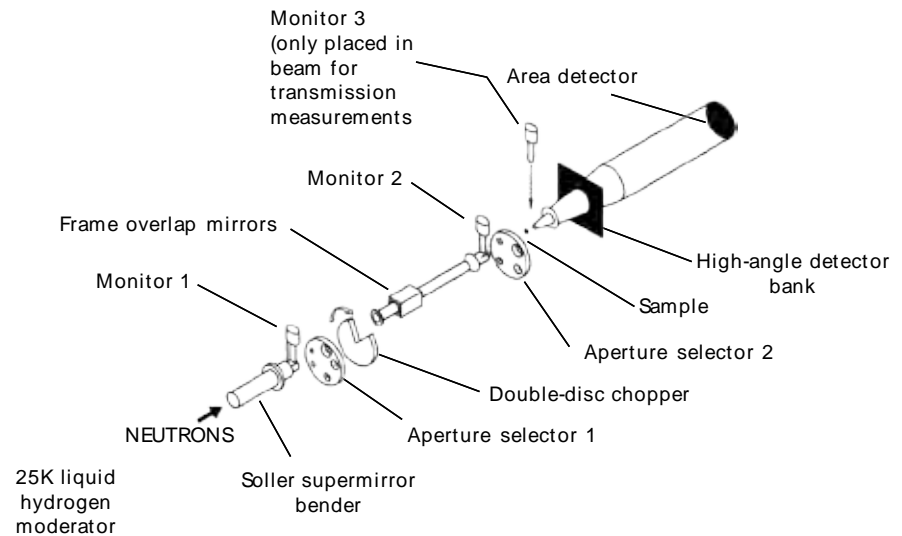
Structure factor: inter-particle information.
Depends on the type of interactions in the system. $S(Q) = 1$ for dilute dispersions

'Flat background'. Generally regarded as due to 'incoherent' scattering, often due to hydrogen



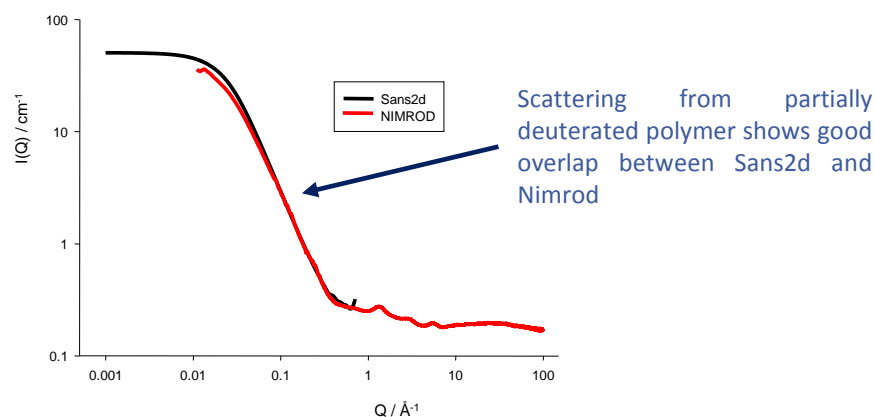
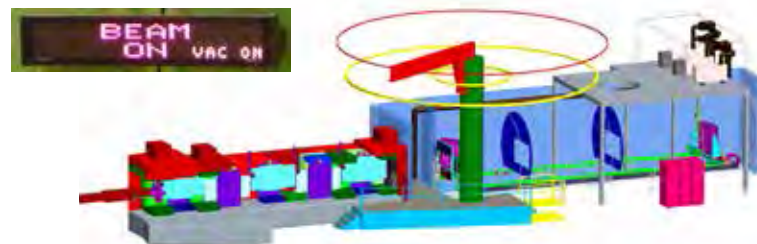
The 1st - Loq

- LOQ was the 1st ISIS SANS instrument and is positioned on the 50 Hz first target station, TS-1
- This instrument has demonstrated the power of SANS at a pulsed source
- Using a fixed sample - main detector distance of 4 m and neutron wavelengths of 2 to 10 Å at 25 Hz a Q-range of 0.007 – 0.3 Å⁻¹ is accessible
- This Q-range can be extended to 1.4 Å⁻¹ by employing the wide-angle detector bank
- Extensive sample environments available
- Sample area is the 'pit' which is accessible via a ladder
- Crane available



The present - Sans2d

- Sans2d is the first SANS instrument to be built on the optimized TS-2
- Operational since 2009
- 2 x 1m² movable detectors provide a uniquely wide simultaneous Q range at good resolution: $Q = 0.001$ to 3 \AA^{-1} , $\lambda = 1.75$ to 17 \AA
- Detector 1 moves +/- 0.3 m sideways, to 12 m from sample. Detector 2 may be offset to 1.4 m sideways in 3.25 m diam vacuum tank and rotates to face sample
- Combining the low background and high flux of Sans2d has allowed weakly scattering samples to be studied more efficiently
- Crane available – shared with Zoom



For further info contact Sarah Rogers: sarah.rogers@stfc.ac.uk



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Larmor: Flexible Spin-Echo for Larmor Precession

- Shutter opened on 20th March 2014
- SANS is available now. The spin-echo setup is being developed with the NWO and TU-Delft over the next year
- Polarized Beam with Analysis
- Wavelengths 0.5 – 13 Å gives SANS Q range of 0.004 - 1.6 Å⁻¹
- Length scales of ~20 nm to 20 μm can be studied using SESANS
- Large block house but detector can move 90° on air pads
- Huber sample stack allows for accurate sample alignment
- Crane available



Future 2 - Zoom

- Currently under construction
- Planned open shutter date is November 2016
- SANS will be available from day 1. Polarized and focusing SANS will be delivered later
- Detector can move to 6 or 9 m from sample
- Vacuum tank can also move 3 m for focusing optics installation
- Wavelengths 1.75 – 16.5 Å gives Q range of 0.002 - 1 Å⁻¹
- Focusing will allow ultra low Q of 0.0003 Å⁻¹ ~ 2 mm
- Huber sample stack allows for accurate sample alignment
- Crane available – shared with Sans2d



For further information contact Ann Terry: ann.terry@stfc.ac.uk



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Polymers in solution for drug coatings

Ionic liquid mixtures

Growth of fibrils

Foams

Nanoparticles in metal alloys

Organic Light Emitting Diodes (OLEDs)

Interaction of polymers with DNA

Orientation of peptide fibrils

Anomalous relaxation behaviour in alcohol-water mixtures

Hydrogen loading

Solution scattering

Flux line lattices

Defects in metals

SANS

Carbon nanotubes

Micellization in CO₂

Templating of nanoparticles with micelles and microemulsions

Colloidal crystals

Surfactant stabilised carbon nanotubes

Interfacial structures of polymers at various interfaces

Exchange in nanoemulsions

Interaction of perfume with micelles

Movement of drugs through and into vesicle bilayers

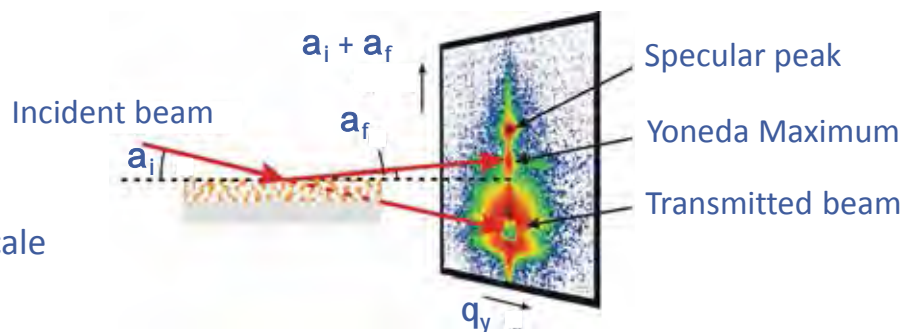
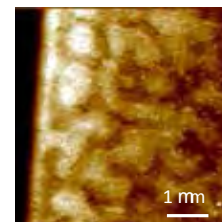
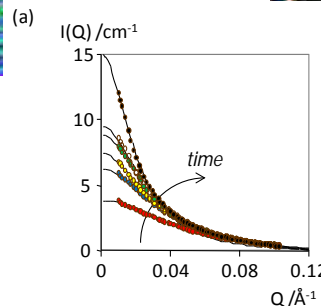
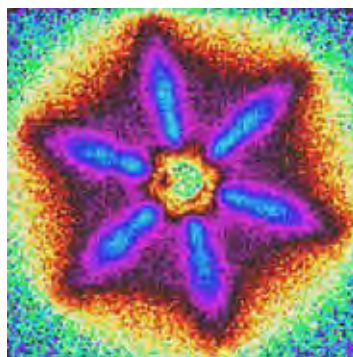


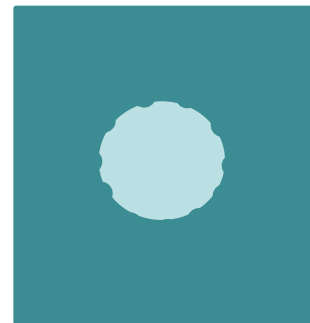
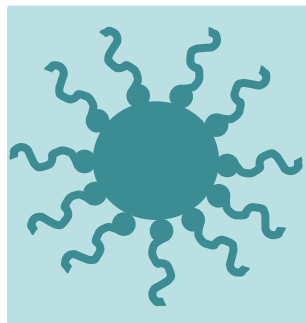
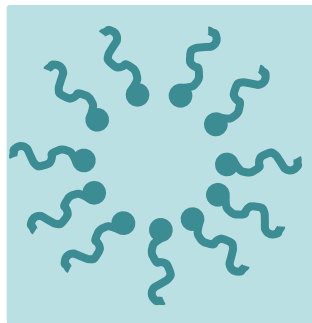
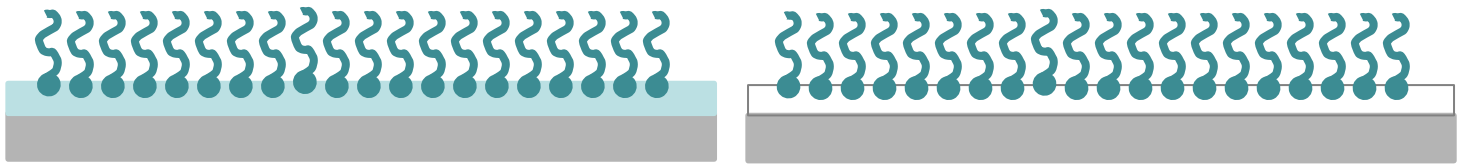
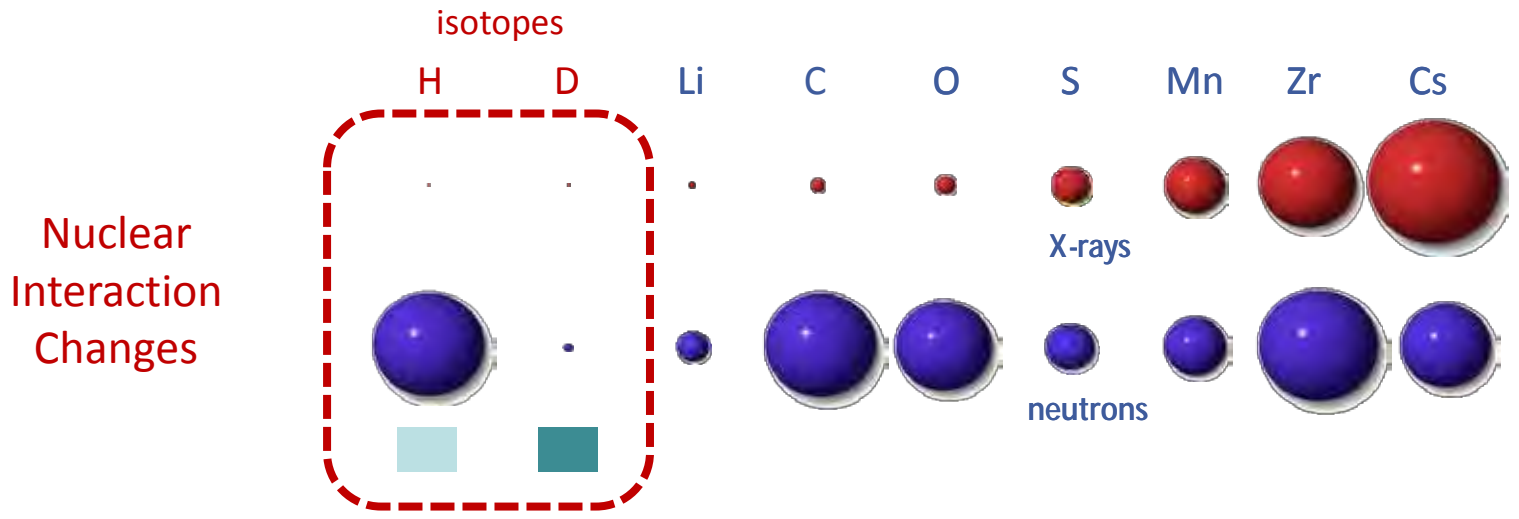
The Sample Environment

Extensive available sample environments allow a broad range of science to be studied via SANS at ISIS.

Sample environment includes:

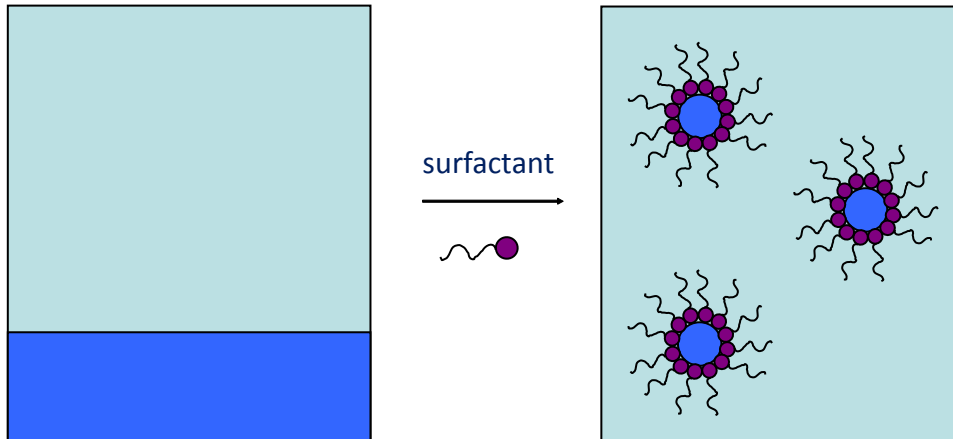
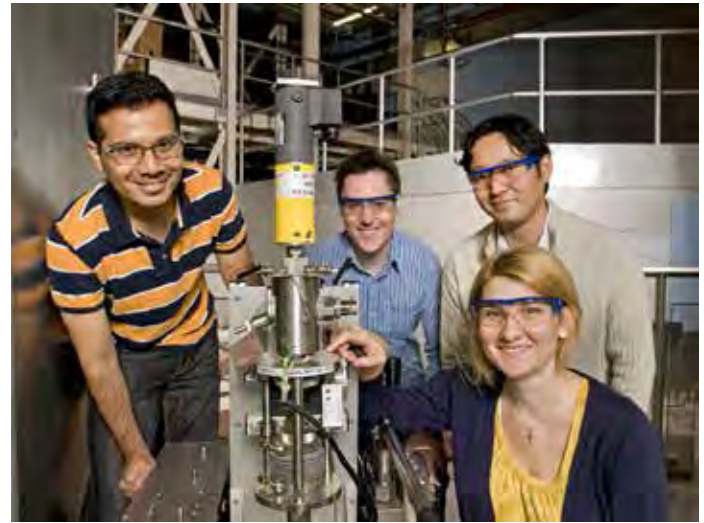
- Standard ISIS cryostats, furnaces and magnets
- Sample changer with temperature control
- Linkham stages for advanced temperature control
- Rheometer and shear cells
- Pressure cell – 600 bar with stirring. Predominantly used with CO₂
- T-jump cell – study non-equilibrium phases
- *In-situ* DLS and UV-vis
- Grazing Incidence SANS (GISANS)
 - Ø Study of in-plane structure on the nm lengthscale
- Stopped-flow – mixing kinetics
- Well equipped offline labs allow for further characterization
 - Ø X-ray sets, AFM, BAM, spectrometers





Collaboration between University of Bristol and the ISIS SANS team studying the modification of the physico-chemical properties of $sc\text{-CO}_2$ with surfactants for use in enhanced oil recovery. Low viscosity of CO_2 promotes fingering through porous media rather than a uniform sweep.

Modifiers commonly used in oily solvents are incompatible with CO_2 . Can self assembled custom-made surfactants be used?



Why Neutron and Small Angle Scattering?

High penetrating power of neutrons allows a p-cell with thick windows to be employed

Using D_2O allows us to see 'nanopools' of water in the CO_2

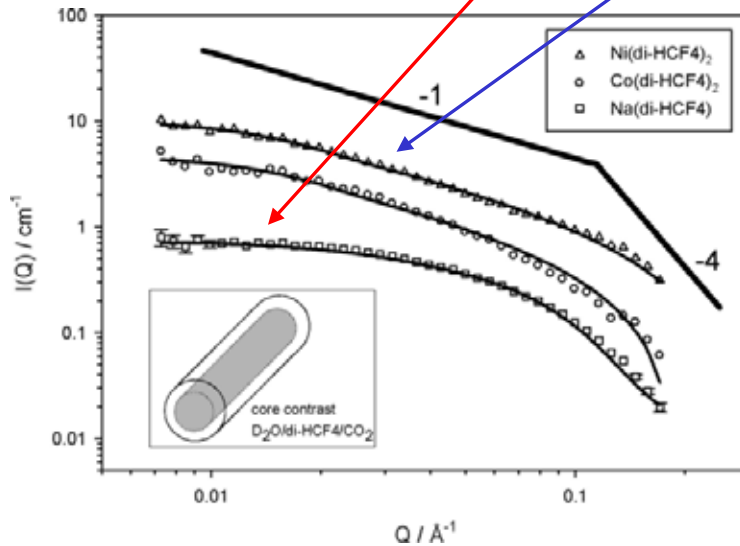
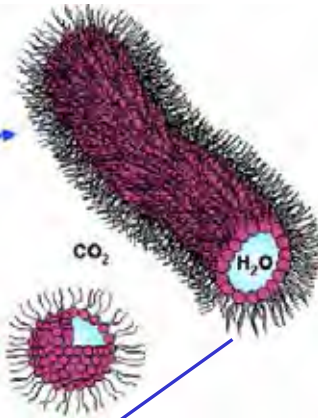
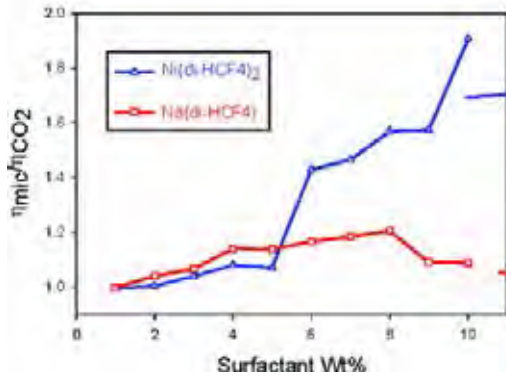
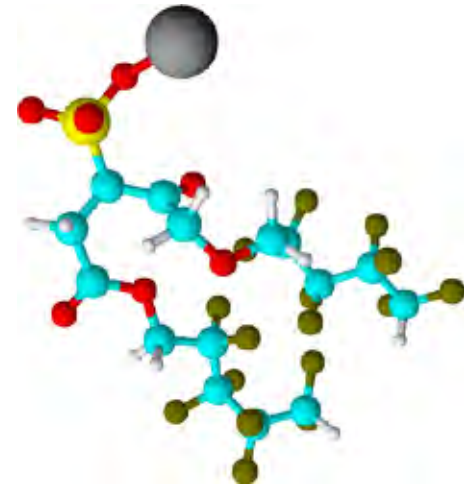
Length-scales being probed are ideal for SANS



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Na⁺ or Co²⁺ or Ni²⁺n(D2O)



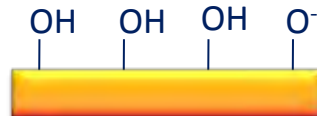
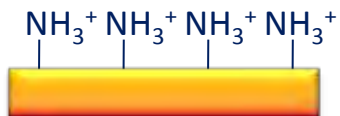
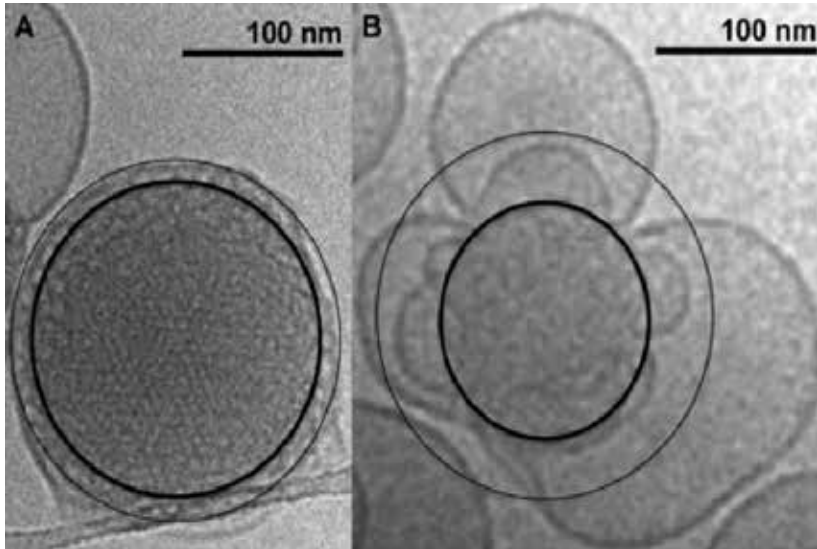
Results

Altering the counterion of the CO₂ active surfactant DHCF4 from Na to Ni or Co causes a viscosity enhancement of up to 90% compared to pure CO₂

Why? Neutrons have the answer! Micelle shape changes from spherical to wormlike as counterion changes from Na⁺ to Co²⁺ or Ni²⁺

Collaboration between Lund University and the ISIS LSS team studying reversed lipid liquid crystalline nanoparticles as drug delivery vehicles

These systems are stabilised by low fractions of the surfactant P80. Location of the P80 within the particle largely control their function



Why SAS?

Neutrons are non-destructive so samples are not altered by beam damage

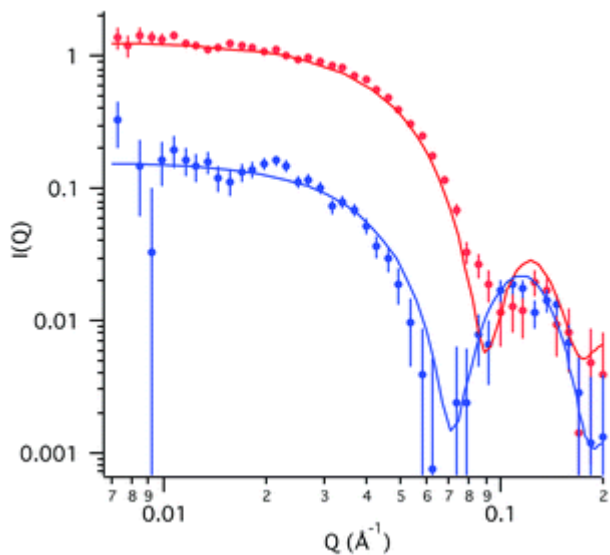
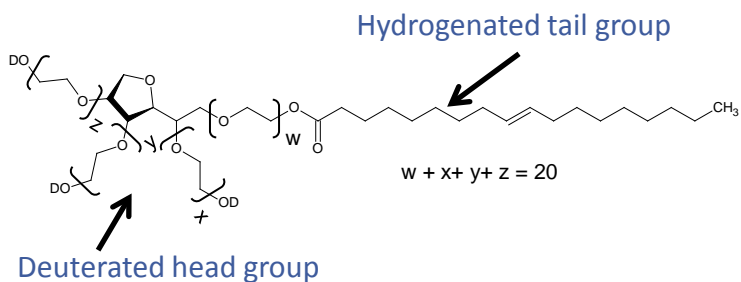
Contrast variation can be used to highlight specific parts of the system

Length-scales being probed are ideal for SAS



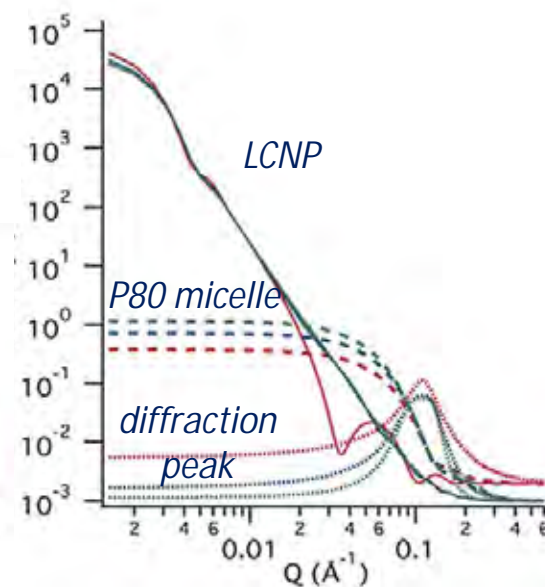
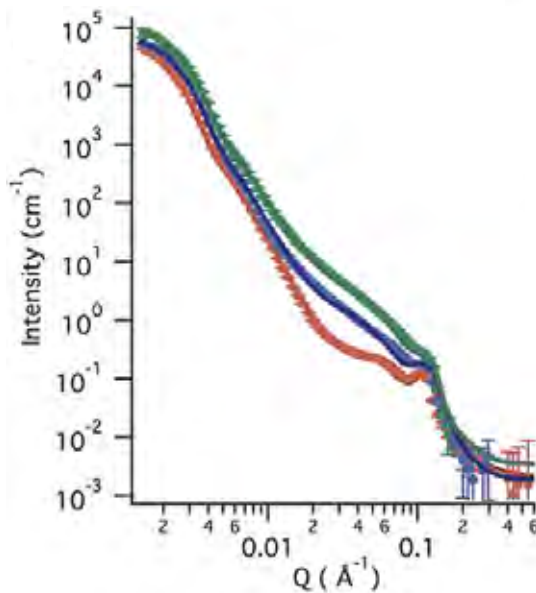
Results

SANS reveals the core-shell structure of the d-P80 micelles



SANS profiles in P80 micelles in H₂O (red) and 1:1 H₂O:D₂O (blue)

SANS profiles with modelled contributions shown

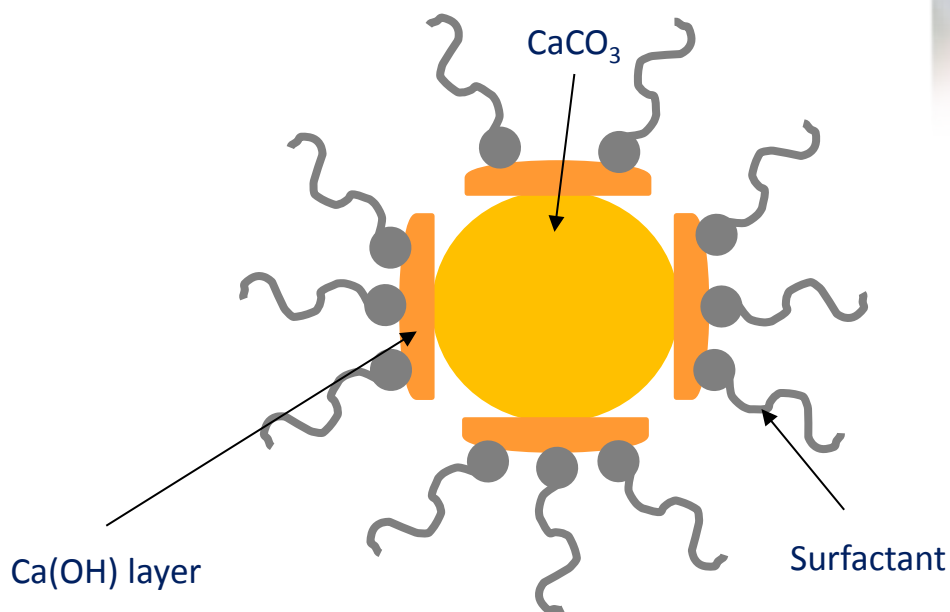


Results

CV-SANS, CV-NR and SAXS have been used to determine a detailed picture of the internal structure of the LCNPs, where the P80 is located within these structures and how the particles are stabilised

Work carried out by Infineum studying engine oil additives which consist of calcium carbonate nanoparticles – CaCO_3 - stabilized by a sulfonate surfactant. The stability of these particles is crucial for their correct performance.

The combustion process can produce a considerable amount of water: how does the presence of water effect these particles?



Why Neutron and Small Angle Scattering?
H/D Contrast provides direct view of water
Length-scales being probes are ideal for SANS



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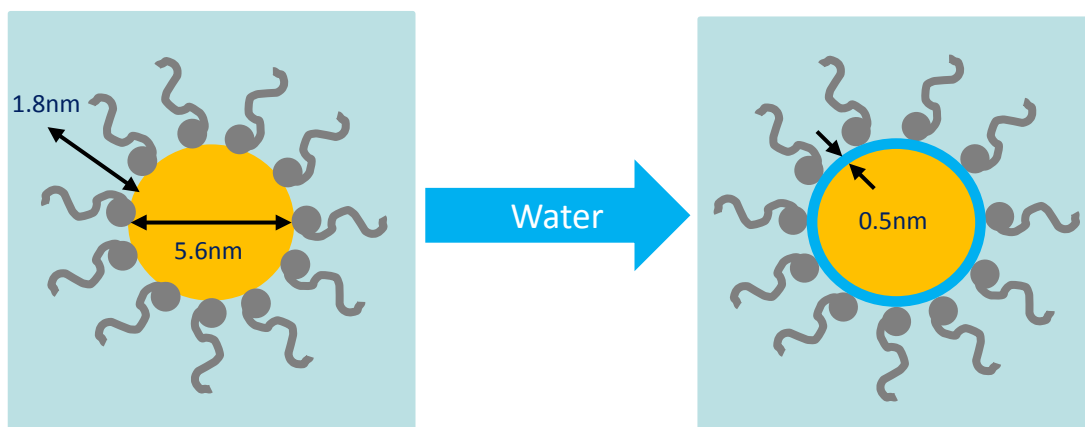
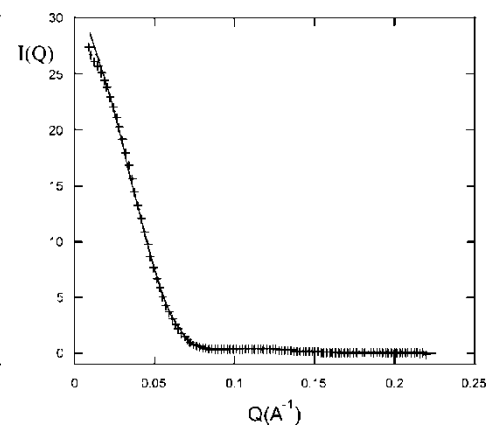
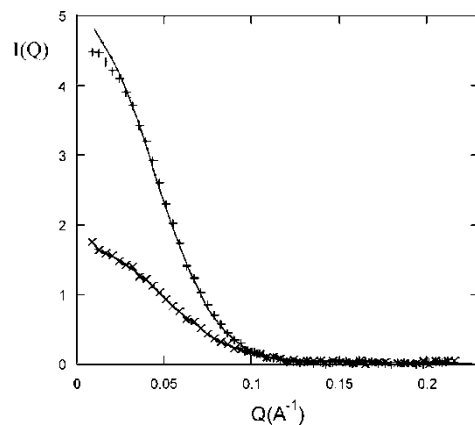
ISIS

Results

CaCO₃ particles are spherical with dia. ~5.6nm

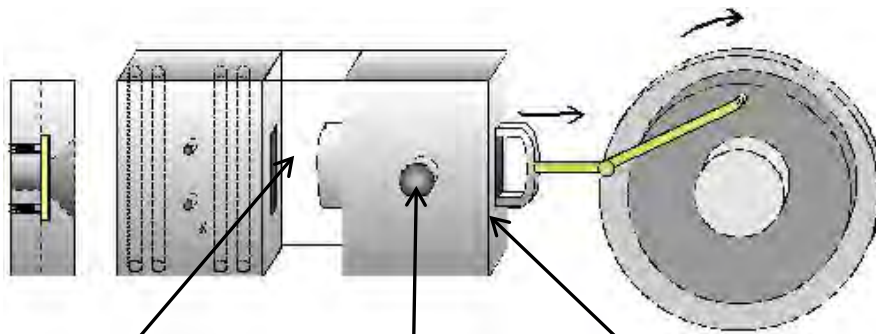
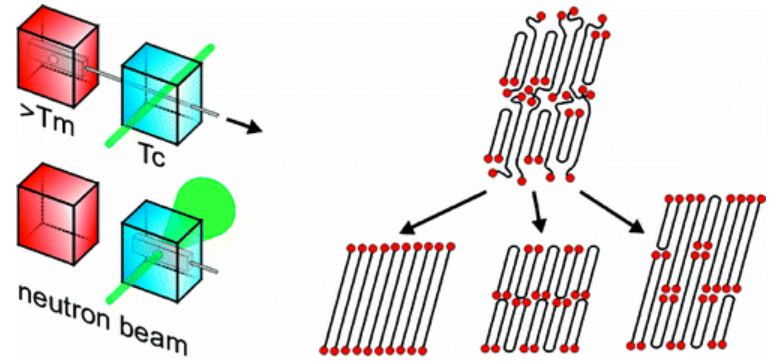
Surfactant monolayer is of thickness ~1.8nm

Water layer inserts between the calcium cation at the surface of the particle and the sulfonate anion



Collaboration between University of Sheffield and the ISIS SANS Team studying transient phases in polymer crystallisation using a temperature jump (T-jump) cell designed for SAS beamlines.

Polymer crystallisation is a highly non-equilibrium process and several different lamellar structures are possible



Sample sits here for ambient conditions

Neutrons

Sample is moved to the heated block for the SANS measurement

Why SANS?

SAS is a powerful technique for studying lamellar structures

Using selectively deuterated segments, SANS can provide information on the location and state of order of such segments



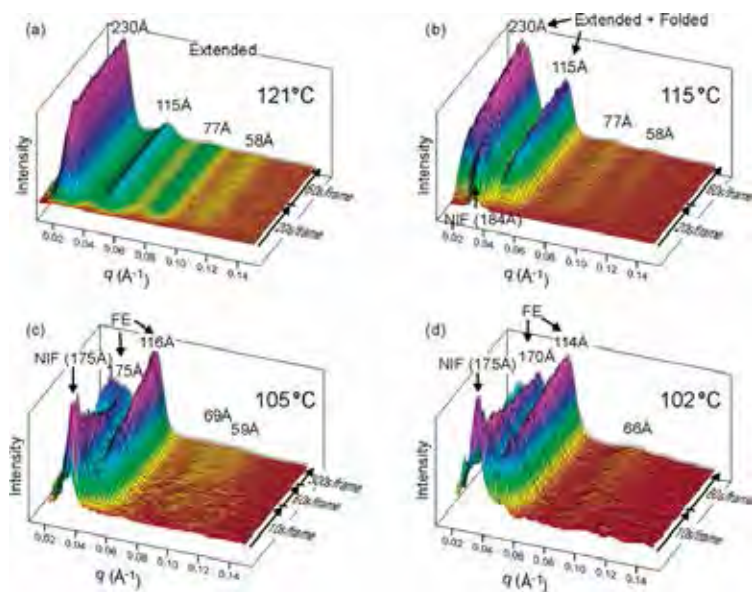
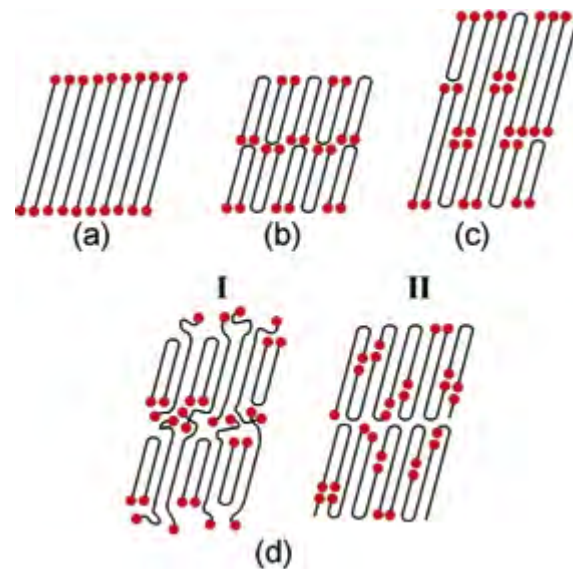
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Results

Material $C_{12}D_{25}C_{192}H_{384}CHDC_{11}D_{23}$ used

Lamellar structures possible are (a) extended chain form, (b) once-folded chain form, (c) triple-layer mixed folded-extended (FE) form and (d) alternative models for the noninteger folded (NIF) form



Results

NIF form has a lifetime of ~ 1 minute – time resolution achievable via SANS

Real-time SANS ‘snap shots’ reveal structural changes with time and temperature

Thanks for listening!

Questions?



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